



Tetrahedron Letters 44 (2003) 2683-2685

TETRAHEDRON LETTERS

An efficient asymmetric synthesis of Fmoc-L-cyclopentylglycine via diastereoselective alkylation of glycine enolate equivalent

Satendra Singh* and Michael W. Pennington

BACHEM Bioscience Inc., 3700 Horizon Drive, King of Prussia, PA 19406, USA Received 17 December 2002; revised 4 February 2003; accepted 6 February 2003

Abstract—Stereoselective alkylation of the enolate derived from benzyl (2R,3S)-(-)-6-oxo-2,3-diphenyl-4-morpholinecarboxylate (1) with cyclopentyl iodide afforded anti- α -monosubstituted product, benzyl (2R,3S,5S)-(-)-6-oxo-2,3-diphenyl-5-cyclopentyl-4-morpholinecarboxylate (3) in 60% yield. Catalytic hydrogenolysis over PdCl₂ cleaved the auxiliary ring system to give L-cyclopentylglycine (4) in 84% yield. Subsequent protection of the α -amino function with Fmoc-OSu gave Fmoc-L-cyclopentyl-glycine (5) in high yield. © 2003 Elsevier Science Ltd. All rights reserved.

1. Introduction

Nature has afforded us 22 naturally occurring coded amino acids commonly found in proteins isolated from eukaryotic and prokaryotic sources. A host of enzymes are present in nature to extend this repertoire much further by utilizing post-translational modification.¹ A variety of synthetic means have also been established to produce a plethora of non-proteinogenic amino acids as tools to investigate enzymatic mechanisms, extend biological half-life, establish a specific conformational determinant or increase potency of therapeutically interesting peptides.² Of these, stereoselective homologation of readily available chiral auxiliaries, such as cyclic glycine enolate equivalents derived from bis-lactim ethers,³ imidazolidinones,⁴ and oxazinones⁵ are par-Besides, glycylsultam⁶ ticularly useful. and pseudoephedrine glycinate hydrate⁷ are also useful α amino acid templates.

Cyclopentylglycine (Cpg) is a competitive inhibitor of isoleucine uptake in *E. coli*⁸ and also has been used in designing angiotensin II antagonists.⁹ It has been synthesized via $S_N 2$ displacement of bromoglycinate with an organometallic reagent followed by epimerization.¹⁰ Syntheses of racemic 2-cyclopentenylglycine,¹¹ cyclopentylglycine,⁸ and 2-cyclopentadieneylglycine¹² have also been reported. In this publication, we wish to report a short and efficient asymmetric synthesis of

Fmoc-L-cyclopentylglycine (5) by using benzyl (2R,3S)-(-)-6-oxo-2,3-diphenyl-4-morpholinecarboxylate (1) as a template. The reasons behind choosing this chiral auxiliary were: (1) commercial availability, (2) excellent optical purity of the final product, (3) high reactivity towards unactivated electrophiles and (4) scalability.

2. Results and discussion

As shown in Scheme 1, chiral auxiliary 1 was alkylated with cyclopentyl iodide in the presence of lithium bis(trimethylsilyl)amide base. Enolate generation at -78° C followed by quenching with alkylating agent at the same temperature did not result in any reaction. Optimum conditions utilized dissolving 1 and cyclopentyl iodide in THF/HMPA (10:1) by heating to $\sim 35^{\circ}$ C, generating enolate at -78° C and allowing the reaction mixture to warm to room temperature over a period of 2 h. Under these conditions, alkylated product 3 was obtained in 60% yield.[†]

Variations in experimental conditions, such as longer reaction time, increasing the amount of base (>1.5

Keywords: Fmoc-L-cyclopentylglycine; cyclopentylglycine; glycine enolate equivalent.

^{*} Corresponding author. Tel.: +1-610-239-0300; fax: +1-610-239-0800; e-mail: ssingh@usbachem.c

[†] **Compound 3**: ESI-MS: 456 ($C_{29}H_{30}NO_4$) (M+H)⁺. ¹H NMR (600 MHz, CDCl₃): δ 7.26–7.07 (13H, m), 6.52 (2H, d, J=6.8 Hz), 6.00 (1H, d, J=3.0 Hz), 5.13 (1H, d, J=3.0 Hz), 4.91 (1H, d, J=12.0 Hz), 4.85 (1H, d, J=12.0 Hz), 3.74 (1H, m), 2.44 (1H, m), 1.90–1.80 (4H, m), 1.65–1.55 (4H, m); mp 216–217°C; $[\alpha]_{24}^{24}$ –44.88° (c 0.5, CH₂Cl₂). Anal. (recrystallized from EtOAc/hexanes) calcd for C₂₉H₂₉NO₄: C, 76.48; H, 6.37; N, 3.07. Found: C, 76.58, H, 6.29; N, 3.28.



Scheme 1.

equiv.) or cyclopentyl iodide (>5 equiv.) and substituting the solvating agent HMPA with DMPU either did not result in any improvements or lowered the yield of the alkylation product **3**. Since cyclopentyl iodide is an unactivated electrophile, raising the reaction temperature to 40°C was expected to improve the yield of alkylation product **3**. However, reaction at 40°C for 1 h resulted in the formation of 10–15% of side product **6** (Fig. 1), in addition to alkylation product **3**.

It is important to note that no dialkylation product was formed. Furthermore, only *trans*-alkylated product (3) formed and no *cis*-diastereoisomer (i.e. 2R,3S,5R) was detected by HPLC. The high diastereoselectivity of the enolate alkylation can be explained by considering the expected boat conformation of **2** that disposes the phenyl ring at C-3 in a pseudoaxial orientation, creating steric shielding of the same face at C-5 position from electrophilic attack as shown in Scheme 1.

No purification by column chromatography was necessary. The *trans*-alkylated product **3** was easily crystallized out as a white solid from EtOAc/hexane in 60% of isolated yield (>99% de by HPLC) after standard workup (cf. *syn*-alkylated product is usually an oil).¹⁰ Additional evidence of stereoselectivity was apparent from ¹H NMR as the methine protons (CH) at C-2 and C-3 appeared at δ 6.00 and 5.13 ppm, respectively (0.87 ppm apart, characteristic of *anti*-alkylation product; cf. 0.6–0.7 ppm apart for the *syn*-alkylated product).¹⁰

Cleavage of the auxiliary ring system 3 was performed using H_2 and PdCl₂ as a catalyst in THF/MeOH (2:1) solvent mixture at 60 psi for 48–60 h. After removing the catalyst (pyrophoric), the solvent was removed and the residue was triturated with ether to afford Cpg 4 in almost quantitative yield. The contaminating byproduct, 1,2-diphenylethane was easily removed by extracting the dilute aqueous HCl solution of 4 with EtOAc. Removal of aqueous solvent followed by crystallization of the syrup from MeOH/EtOAc afforded 4 in 84% yield.^{\ddagger}

Protection of the α -amino function of Cpg **4** was accomplished by treating it with Fmoc-OSu in the presence of Na₂CO₃ overnight in dioxane/H₂O (1.5:1) solvent mixture. Fmoc-Cpg-OH (**5**) was obtained in quantitative yield after crystallization from EtOAc/hexane.

Thus, an efficient asymmetric synthesis of Fmoc-Cpg-OH (5) from a commercially available chiral auxiliary was successfully accomplished on multigram scale. Synthesis was easy to perform, as no chromatographic purification was required at any step. Furthermore, excellent optical purity (>99%) and high yield (50% overall) was obtained.



Figure 1.

[‡] **Compound 4:** ESI-MS: 144 ($C_7H_{13}NO_2$) (M+H)⁺. ¹H NMR (600 MHz, DMSO-*d*⁶): δ 4.74 (1H, m), 2.48 (1H, m), 1.78–1.69 (2H, m), 1.53–1.42 (4H, m), 1.39–1.25 (2H, m); mp 245–247°C (dec.); $[\alpha]_{D}^{24}$ +12.34° (*c* 0.5, MeOH); (lit.¹⁰, +11.6°, *c* 0.49, 1N HCl)). Anal. (recrystallized from MeOH/EtOAc) calcd for $C_7H_{13}NO_2$: C, 58.74; H, 9.09; N, 9.79. Found: C, 58.70, H, 8.85; N, 9.99.

References

- 1. Atkins, J. F.; Gesteland, R. Science 2002, 296, 1409– 1410.
- (a) O'Donnell, M. J. (Ed.). α-Amino Acid Synthesis (Tetrahedron Symposium-in-Print). Tetrahedron 1988, 44, 5253–5614; (b) Williams, R. M. The Synthesis of Optically Active α-Amino Acids; Pergamon Press: Oxford, 1989; Vol. 7; (c) Duthaler, R. O. Tetrahedron 1994, 50, 1539–1650; (d) Seebach, D.; Sting, A. R.; Hoffmann, M. Angew. Chem., Int. Ed. Engl. 1996, 35, 2708.
- 3. Schoellkopf, U. Tetrahedron 1983, 39, 2085–2091.
- Seebach, D.; Juaristi, E.; Miller, D. D.; Schickli, C.; Weber, T. Helv. Chim. Acta 1987, 70, 237–261.
- 5. Williams, R. M.; Im, M.-N. J. Am. Chem. Soc. 1991,

113, 9267–9286.

- 6. Oppolzer, W.; Moretti, R.; Zhou, C. Helv. Chim. Acta 1994, 77, 2363–2380.
- Meyers, A. G.; Schnider, P.; Kwon, S.; Kung, D. W. J. Org. Chem. 1999, 64, 3322–3327.
- Harding, W. M.; Shire, W. J. J. Biol. Chem. 1954, 206, 401–410.
- Nyeki, O.; Szalay, K. S.; Kisfaludy, L.; Karpati, E.; Szporny, L.; Makara, G. B.; Varga, B. J. Med. Chem. 1987, 30, 1719–1724.
- Williams, R. M.; Sinclair, P. J.; Zhai, D.; Chen, D. J. Am. Chem. Soc. 1988, 110, 1547–1557.
- Dennis, R. L.; Plant, W. J.; Skinner, C. G.; Southerland, G. L.; Shive, W. J. Am. Chem. Soc. 1955, 77, 2362–2364.
- 12. Dialer, H.; Steglich, W.; Beck, W. Tetrahedron 2001, 57, 4855–4861.